# GEOCHEMICAL STUDIES - 3 GEOCHEMISTRY OF SOME ALBERTA SHALES AND ASSOCIATED KEROGEN

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## "GEOCHEMICAL STUDIES"

This series of reports, under the general title "Geochemical Studies", makes available geochemical data on the formation fluids and rocks of Alberta which would otherwise remain unpublished. Two types of data fall in this category. First, "Geochemical Studies" will act effectively as a document depository in cases where a formal publication is available, but without the raw data having been published; additional interpretations may be included if pertinent. Second, "Geochemical Studies" will include both the raw data and a minimal descriptive report in the case where no formal publication is planned.

# **CONTENTS**

Abstract	••••••	
Introduction	•••••	2
Analytical prod	cedures	2
	en	
Geochemistry	of Alberta shales	Ę
	ge bulk composition	
Trace	element geochemistry	ξ
	Introduction	ξ
	Alkali metals - sodium	
	Alkaline earth metals - magnesium, calcium and strontium 1	
	Zinc and cadmium	
	Vanadium	
	Chromium	
	Manganese	
	Copper and silver	
	Molybdenum	
On a share inter-	Lead	
	of kerogen	
	ntal composition	
	roperties	
	IR properties	
	nents	
References		3.
	V 2: U	
Appendix A.	Geochemistry of Alberta shales and associated kerogen	
	- Data sheets A	1
Appendix B.	Organic petrography and reflectance measurements of	
	selected Alberta shales (report by INRS) B	1
Figure 1.	Distribution of Alberta shales studied. Solid	
	tie line joins samples from the Viking Formation,	
	and dashed tie line joins samples from the non-oilsands	
		3
Figure 2.	Relations of average values of [Ca+Mg+Srx10 = 100]	
	in shales and formation waters from 3 groups	1
Figure 3.	A. van Krevelan diagram (atomic H/C vs atomic O/C)	•
90.0 0.	showing maturation regions; B. van Krevelen diagram	
	for kerogens from the Alberta Basin. The cross (+)	
	ior rerogens from the Alberta Dasill. The Closs (T)	

	is sample K-16 from the Middle Devonian Nahanni Formation of northeast British Columbia (4125 m, R <sub>o</sub> ~ 2.5, estimated present reservoir temperature 180°C). C. van Krevelan diagram for kerogens from the Alberta Basin showing the general environment of deposition of the shales from which the kerogen was extracted, and the estimated maximum temperatures (°C). Solid tie line - Viking Formation;
Figure 4.	dashed tie line - non-oilsands Mannville Group
Figure 5.	Relation of: A. ESR broad band spin concentration (spins/gx10 <sup>18</sup> ); B. ESR g-value; and C. ESR broad band width (mT) to estimated maximum temperature (°C) in kerogens from the Alberta Basin. Solid tie line - Viking Formation; dashed tie line - non-oilsands
Figure 6.	Mannville Group
Figure 7.	Relation of aliphatic-carbon/aromatic-carbon ratio and estimated maximum temperature (°C) in kerogens from the Alberta Basin. Solid tie line - Viking Formation; dashed tie line - non-oilsands Mannville Group
Table 1.	Comparison of the proportions of shale and shale-samples in selected stratigraphic units. Alberta Basin shale volumetric data from Hitchon (1968)
Table 2.	Comparison of ESR parameters on kerogen separated from Toarcian shales, Paris Basin
Table 3.	Comparison of the composition (%) of non-calcareous Alberta Basin shales with average geosynclinal and platform shales (from Wedephol, 1969, Table 8-4)
Table 4. Table 5.	Estimated maximum burial depth and temperature
	see Appendix A

## **ABSTRACT**

A set of 28 shales from the Alberta Basin, ranging in age from Late Cretaceous to Middle Cambrian, was studied with respect to their major and trace element composition, and the kerogen extracted from 25 of them for detailed analysis. This Open File report contains the basic data on the shales and kerogen. It is not intended to report this information elsewhere.

Although the shales are widely scattered areally, on a volumetric basis the Upper Cretaceous is underrepresented and the Lower Cretaceous overrepresented. The average for 24 non-calcareous shales from the Alberta Basin is close to the average shale in geosynclines. A decrease in SiO2 content with age suggests a decrease in silt (quartz) content rather than a change in clay mineralogy. General statistics are provided for groups of shales based on the three crude oil families distinguished by previous authors i.e. Group 1: U. Cretaceous Colorado and post-Colorado reservoirs; Group 2: L. Cretaceous Mannville Group, Jurassic, Carboniferous and U. Devonian Wabamun Group reservoirs; Group 3: U. Devonian Winterburn and Woodbend Groups and Beaverhill Lake Formation reservoirs. For Cd, Cr, Mn, Fe, Co, Ni and Mo there is no significant difference among the three groups of shales. Sodium in formation water and crude oil increases with depth. whereas, for reasons not clear, Na in shales decreases. The sharp decrease in the contents of Zn, Cu and Pb with depth is associated with the increased salinity (chloride content) of the associated formation waters and may relate to the greater ease of removal of these elements from shales as chloride complexes of the metals.

The majority of the kerogen samples are Type III, the rest being Type II; with one exception they are low-S types. Using estimated maximum temperature from a previous study, it is suggested that there is a tendency for loss of both O and S with increased temperature and therefore a relative increase in N as a residual effect. With respect to the ESR properties of the kerogen, increased temperature results in increased spin concentrations, lower g-values, and wider lines. At any specific temperature, there is a wide range in the NMR aliphatic-carbon/aromatic-carbon ratio.

## INTRODUCTION

The Alberta Basin is well explored, with abundant geological and geochemical data. Comprehensive reports are available on the geochemistry of crude oils and of hydrocarbons extracted from potential source rocks, but there is only one published study (McIver, 1967) on the geochemistry of kerogen extracted from shales. Accordingly, a suite of 28 shales was collected in the late 1970s from several sources, in an attempt to represent major argillaceous units both stratigraphically and areally (Figure 1). However, when the relative volumes of shales in the main stratigraphic units are compared with the proportion of samples from the same unit (Table 1) the Upper Cretaceous is underrepresented and the Lower Cretaceous overrepresented. Most of the samples are from core stored by the Alberta Energy Resources Conservation Board; two samples (RCA 65-1, RCA 67-1) are from core from shallow test holes drilled by the Alberta Research Council.

The purpose of this Open File report is to present (Appendix A) details of the stratigraphy, location, depth, sample description and depositional environment (determined from geological information) of these 28 samples, together with inorganic and organic determinations. Properties of the associated kerogen are included if the kerogen was extracted. There is a brief description of the analytical techniques but minimal interpretation of the data. This Open File report contains the only summary of the geochemistry of these shales and their associated kerogen.

## **ANALYTICAL PROCEDURES**

## SHALES

Samples were chipped clean and crushed in a shatterbox to pass 325 mesh. The partial major element analysis carried out by Barringer Research Limited (Rexdale, Ontario) has been converted to oxide percentages in the data tables in Appendix A. Twelve trace elements (Ag, Cd, Co, Cr, Cu, Mn, Mo, Ni, Pb, Sr, V, Zn) were also determined by Barringer Research Limited using a multielement radio frequency emission spectrometer following lithium metaborate fusion. Detection limits (ppm) are as follows: Ag 0.004; Cd 0.002; Co 0.003; Cr 0.001; Cu 0.001; Mn 0.001; Mo 0.005; Ni 0.006; Pb 0.008; Sr 0.00002; V 0.006; Zn 0.002 (these detection limits refer to pure solutions).

Organic C and extractable organics were determined by the Institute of Sedimentary and Petroleum Geology, Geological Survey of Canada (Calgary, Alberta). Vitrinite reflectance and organic matter petrography determinations (Appendix B) were carried out on selected samples by Institut Nationale de la Recherche Scientifique, Universite du Quebec (Ste. Foy, Quebec).

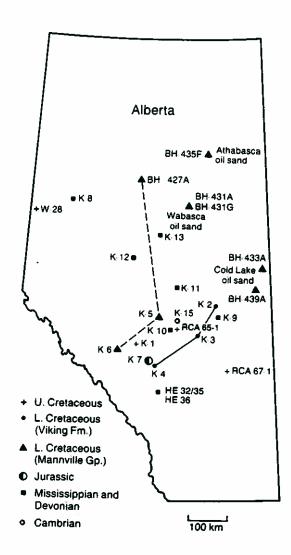


Figure 1. Distribution of Alberta shales studied. Solid tie line joins samples from the Viking Formation, and dashed tie line joins samples from the non-oilsands Mannville Group.

Table 1. Comparison of the proportions of shale and shale-samples in selected stratigraphic units. Alberta Basin shale volumetric data from Hitchon (1968).

Stratigraphic Unit		S	Samples studied		
	% Total basin	% in unit	% Total shale in basin	No.	% Total
Tertiary	1.3	75.0	1.8	-	-
U. Cretaceous	27.3	92.6	44.3	4	14
L. Cretaceous	13.8	73.1	17.8	11	39
Jurassic	2.4	57.7	2.4	1	3.5
Triassic	2.0	75.7	2.6	-	-
Permian	0.4	15.3	0.1	-	-
Carboniferous	9.0	35.5	5.6	3	11
U. Devonian	22.0	36.4	14.1	5	18
M. Devonian	9.0	15.7	2.5	3	11
Silurian	1.3	5.0	0.1	-	-
Ordovician	3.3	15.0	0.9	-	-
Cambrian	8.2	54.1	7.8	1	3.5

#### KEROGEN

Kerogen was separated using the techniques described by Hitchon et al. (1976), in which digestions were held at 40°C and took place under  $N_2$  to avoid an increase in the number of free radicals in the kerogen as a result of conventional hot HF digestion. After digestion in 2N HCl, the residue was centrifuged and washed with hot water until free of Ca ions. Following digestion with 48% HF, a few millilitres of 2N HCl were added to the solution, which was then diluted with a saturated boric acid solution, centrifuged, and washed with hot water. The residue was then centrifuged in a mixture of bromoform and acetone (6:1) to separate the kerogen from pyrite. X-ray diffraction showed K-fluosilicate in all kerogen samples, which was entirely removed with copious volumes of hot water. Only three samples (RCA 65-1, K-1, BH-439A) contained significant amounts of ralstonite, which was quantitatively extracted with 6N HCl at 70°C and analyzed (Hitchon et al., 1976).

Elemental composition was determined by Alfred Bernardt Mikroanalytisches Laboratorium (Elbach uber Engelskirchen, Germany), with O determined directly. The elemental composition in Appendix A is normalized (Sum CHONS=100) and adjusted for pyrite-S (%Fe x 1.1482). Although it was difficult to remove the entrained pyrite

completely (Fe 1 to 22%, mean 8.5%) clean-up of the pyrite separates for S isotope determination was relatively successful; C in 13 pyrite concentrates was in the range 1.3 to 11.8% (average 4.9%).

The ESR properties were determined by Dr. K.F. Schulz, Alberta Research Council on a Varian Model 4500 ESR spectrometer, operating at a nominal frequency of 9.3 GHz, with power levels maintained low enough to avoid saturation effects. Although samples were run in air and *in vacuo* only the latter results are reported here. Corrections were made for the pyrite content (range 1.97-47.47%; average 18.5%). A check on the technique used was made by analysing kerogen previously analysed by the Institut Francais du Petrole (B. Durand, personal communication); Table 2 shows that the results are in close agreement.

The solid state <sup>13</sup>C NMR spectra were obtained at 22.6 MHz on a Bruker CXP180/90 NMR spectrometer using the cross polarization (CP) technique with magic angle spinning (MAS). A contact time of 1 ms and a repetition time of 2 s were selected. The MAS speeds were 3-3.4 KHz. Chemical shifts were determined by substitution relative to tetramethylsailane (TMS). Each spectrum was the Fourier transform of 3000-10,000 free-induction decay curves. The <sup>13</sup>C NMR spectra were run by Dr. John Ripmeester (Division of Chemistry, National Research Council of Canada, Ottawa, Ontario).

Sulfur isotopes were determined by Professor H.R. Krouse (The University of Calgary) on the kerogen, pyrite and extractable organic matter on selected samples, depending on sample size available.

#### GEOCHEMISTRY OF ALBERTA SHALES

#### AVERAGE BULK COMPOSITION

The average of 24 non-calcareous shales from Appendix A is close to that for the average shale in geosynclines (Table 3). The four samples not included in Table 3 had either anomalous total oxides (RL 2/4, K-16) or were calcareous (HE 32/35, K-12) with indeterminate CaO content (>9.2%). Individual major oxides, however, show wide concentration ranges (see Table 3). In part, these wide concentration ranges are a reflection of the wide ranges of depositional environments, diagenesis and hence mineralogy of shales in the Alberta Basin. For example, many analyses reporting SiO<sub>2</sub>>60% are from shales classified as 'silty', although it is recognized that this classification is highly subjective without a grain size analysis. The highest SiO<sub>2</sub> (77%, RCA 67-1) is from a shale reported as

Table 2. Comparison of ESR parameters on kerogen separated from Toarcian shales, Paris Basin.

(1)		Alberta Research Council	Institut Francais du Petrole
Ancerville	Spin No. (x10 <sup>18</sup> )	1.2	<del>-</del>
	g-value	2.00285	•
	Line width (mT)	5.7	•
Bouchy	Spin No. (x10 <sup>18</sup> )	4.1	4.7
	g-value	2.00287	2.0028
	Line width (mT)	6.3	6.4
Essises	Spin No. (x10 <sup>18</sup> )	1.4	2.5
	g-value	2.00278	2.0029
	Line width (mT)	6.1	6.1
Cesarville	Spin No. (x10 <sup>18</sup> )	1.2	1.9
	g-value	2.00274	-
	Line width (mT)	7.2	5.1
Vacherauville	Spin No. (x10 <sup>17</sup> )	3.8	5.2
	g-value	2.00300	-
481	Line width (mT)	6.0	5.7
Joinville	Spin No. (x10 <sup>17</sup> )	5.6	6.2
	g-value	2.00269	•
	Line width (mT)	8.3	6.7
Dontrien	Spin No. (x10 <sup>17</sup> )	0.7	1.2
	g-value	2.00281	-
	Line width (mT)	8.2	5.6

Table 3. Comparison of the composition (%) of non-calcareous Alberta Basin shales with average geosynclinal and platform shales (from Wedepohl, 1969, Table 8-4).

	Alberta Basin (n=24)*		Shales (geosynclines)	Shales (platforms)	
	Average	Range	(n=277)	(n=6800)	
SiO <sub>2</sub>	58.4	32-77	58.9	50.7	
TiO <sub>2</sub>	0.6	0.3-0.9	0.78	0.78	
Al <sub>2</sub> O <sub>3</sub>	14.8	9-26	16.7	15.1	
Fe as Fe <sub>2</sub> O <sub>3</sub>	5.1	2.9-11.6	6.9	6.7	
MnO	0.04	0.01-0.12	0.09	0.08	
MgO	2.0	0.8-4.7	2.6	3.3	
CaO	1.4	0.2-9.1	2.2	7.2	
Na₂O	0.3	0.0-0.7	1.6	0.8	
K₂O	-	-	3.6	3.5	
H₂O+	-	-	5.0	5.0	
P <sub>2</sub> O <sub>5</sub>	-	-	0.16	0.10	

<sup>\*</sup>Data from Appendix A, except RL 2/4, HE 32/35. K-12 and K-16.

'bentonitic'. Although there are only a few data, there is a suggestion that the  ${\rm SiO_2}$  content of non-calcareous shales in the Alberta Basin decreases with age e.g.

	No.	Avg. SiO₂	Avg. Al <sub>2</sub> O <sub>3</sub>
U. Cretaceous	4	67	14.8
L. Cretaceous (L. Colorado Gp.)	3	65	13.7
L. Cretaceous (Mannville Gp.)	8	57	15.6
Mississippian and Devonian	8	52	14.0

This trend is not reflected in the  $Al_2O_3$  content and might relate to a decrease in silt (quartz) content, rather than a change in clay mineral content.

Other particularly high contents of individual major oxides include 11.6% Fe<sub>2</sub>O<sub>3</sub> and 0.12% MnO in sample BH-427A, which contained pyrite nodules. The range for TiO<sub>2</sub> and Na<sub>2</sub>O is quite narrow.

## TRACE ELEMENT GEOCHEMISTRY

#### Introduction

Based on a very comprehensive study, Deroo et al. (1977) distinguished three crude oil families in Alberta, as follows:

Group 1	U. Cretaceous Colorado and post-Colorado reservoirs
Group 2	L. Cretaceous Mannville Group, Jurassic, Carboniferous and U. Devonian Wabamun Group reservoirs
Group 3	U. Devonian Winterburn and Woodbend Groups and Beaverhill Lake Formation reservoirs

Hitchon and Filby (1983) used simple statistics and an ANOVA, and subsequently the same authors used stepwise multiple-discriminant function analysis and multiresponse permutation procedures (Hitchon and Filby, 1984) to show that trace elements can also be used to classify crude oils into families. All these studies dealt with either the organic (Deroo et al., 1977) or inorganic (Hitchon and Filby, 1983, 1984) parts of crude oils. However, crude oils originate from source rocks, dominantly shales, and there could well be geochemical links between the trace elements in source rocks, those in associated crude oils, and indeed in the associated formation water, the carrier medium for hydrocarbon migration. Hitchon and Filby (1983) pointed out some of these possible links and indicated where future research might be directed. With this latter objective in mind the following notes are intended to point out some interesting simple statistical trends which may have some veracity. However, extreme caution must be used in any extrapolation from such small and varied populations. As appropriate, the data bases used were Appendix A (this report) for shales, Hitchon and Filby (1983) for crude oils and Hitchon et al. (1971) for formation waters. The three groups for each data set were as determined by Deroo et al. (1977).

## Alkali Metals - Sodium

#### General Statistics

Na (ppm or mg/l)	Number of samples reporting element	Minimum	Median	Average	Maximum
GROUP 1			<del> </del>		
Shales (n=7)	7	900	3800	3400	5400
Crude oils (n=20)	20	0.046	0.73	1.54	13.0
Formation waters (n=15)	15	1510	7260	8080	22800
GROUP 2					
Shales (n=12)	12	200	1400	2000	5100
Crude oils (n=25)	22	0.018	0.83	2.94	19.3
Formation waters (n=25)	25	510	10600	15200	42900
GROUP 3					
Shales (n=7)	7	300	900	1300	3500
Crude oils (n=33)	33	0.0174	0.632	4.61	64.7
Formation waters (n=34)	34	9360	48900	48100	92800

## Geochemistry

Although Na, Rb and Cs were determined in the crude oils, and Na, K and Rb in the formation waters, only Na was determined in the shales. The average depths are 950 m, 1270 m and 1960 m for shales from groups 1 to 3, respectively, so the marked increase in the median, average and maximum contents of Na in the formation waters is to be expected given the general increase of salinity with depth. There is a similar trend for the average and maximum contents of Na in the crude oils; this is not, however, a result of entrained formation water (see discussion in Hitchon et al., 1975) but likely related to pH effects and the fact that Na in crude oil probably occurs as the Na salt of a petroleum acid (see discussion in Hitchon and Filby, 1983, pp. 14-16). What is most interesting in the general statistical data is the decrease in the minimum, median, average and maximum for Na in shales, opposite to the trends in the two fluids. At this time there is no obvious explanation.

# Alkaline Earth Metals - Magnesium, Calcium and Strontium

## General Statistics

Mg, Ca, Sr (ppm or mg/l)	Number of samples reporting element	Minimum	Median	Average	Maximum
GROUP 1					
Shales (n=7)	Mg 7	7000	9400	9700	12200
	Ca 7	2000	2700	3900	8300
	Sr 7	62	170	135	183
Formation waters (n=15)	Mg 13	8	46	95	409
	Ca 15	14	72	211	1230
	Sr 15	0.52	20	40	219
GROUP 2		_			
Shales (n=12)	Mg 12	4500	12300	12600	28300
:	Ca 11	1500	4100	9900	31700
	Sr 12	46	160	177	352
Formation waters (n=25)	Mg 25	2	68	370	1760
	Ca 25	12	223	1400	9740
	Sr 25	0.27	26	123	539
GROUP 3					
Shales (n=7)	Mg 7	4400	14100	12400	19800
	Ca 6	16000	22800	28600	65000
	Sr 7	10	182	227	658*
Formation waters (n=34)	Mg 34	428	2230	2320	3990
	Ca 34	1380	17700	17000	38700
	Sr 34	36	644	624	1320

<sup>\*</sup>without this extreme value (>3 $\sigma$  from the mean for all shales) the median, average and maximum values are 172, 156 and 279 ppm, respectively.

# Geochemistry

For the alkaline earth metals, data are available only for shales and formation waters. In the formation waters, there is a general increase in all alkaline earths with depth, and the order is uniformly Ca > Mg > Sr. In the shales the order is generally Mg > Ca > Sr for groups 1 and 2, but is Ca > Mg > Sr for group 3, almost certainly due to the presence of calcite in the shales. A ternary diagram (Figure 2) of average values for [Ca+Mg+Srx10=100] shows that both shales and formation waters have a marked

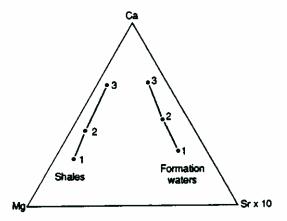


Figure 2. Relations of average values of [Ca+Mg+Srx10=100] in shales and formation waters from 3 groups.

increase in Ca with depth going from group 1 to group 3, mainly at the relative expense of Mg in the case of shales and Sr in the case of formation waters. It is suggested that the effect in shales results essentially from the presence of calcite in increasingly calcareous rocks going from group 1 to group 3 (see trends in Hitchon, 1968, Figures 3 and 4). The trend in the formation waters possibly relates to solubility effects. In a recent study of formation waters from northern Alberta, Hitchon (1993) showed that the control on the content of Sr is by means of different Sr-minerals, depending on the hydrostratigraphic unit -- or more precisely, on parameters such as SO<sub>4</sub> content and dissolved CO<sub>2</sub> species, which generally vary with depth.

#### Zinc and Cadmium

#### General Statistics

Zn, Cd (ppm or mg/l)	Number of samples reporting element	Minimum	Median	Average	Maximum
GROUP 1					
Shales (n=7)	Zn 7	99	161	270	943*
	Cd 7	6.4	13	15	26
Crude oils (n=20)	Zn 20	0.0115	0.270	0.272	0.670
Formation waters (n=15)	Zn 15	0.00	0.08	0.36	3.05
	Chloride 15	798	•	12200	38200
GROUP 2					
Shales (n=12)	Zn 12	65	126	132	261
	Cd 12	10	13	14	22
Crude oils (n=25)	Zn 24	0.0175	0.401	0.456	1.277
Formation waters (n=25)	Zn 25	0.03	0.13	0.32	3.05
	Chloride 25	320	5	25700	95800
GROUP 3					
Shales (n=7)	Zn 7	19	60	61	100
	Cd 7	1.1	14	9	16
Crude oils (n=33)	Zn 28	0.0134	0.140	0.573	5.921
Formation waters (n=34)	Zn 34	0.05	0.16	1.43	27.5
	Chloride 34	15800	•	69700	172000

<sup>\*</sup>without this extreme value (>3σ from the mean for all shales), the median, average and maximum values are 158, 158 and 206 ppm, respectively.

## Geochemistry

Hitchon (1977) discussed the geochemical links between oil fields and ore deposits in sedimentary rocks with special reference to Alberta, and drew an analogy between the features of high salinity and high metal content of geothermal brines and the deeper, hotter, more saline formation waters from Alberta. A similarity was suggested between the leaching of hydrocarbons from shales to produce oil fields and the leaching of shales to produce enhanced contents of metals in formation waters, possibly resulting in ore deposits. This similarity is illustrated by the statistics presented above, in which data on Zn and Cl in formation waters (Hitchon et al., 1971), Zn in crude oils and Zn in shales are grouped according to the crude oil families distinguished by Deroo et al. (1977). In broad terms, the statistics may be summarized as follows:

	Group 1	Group 2	Group 3
Zn in shales	High	Intermediate	Low
CI in formation waters	Low	Intermediate	High
Zn in formation waters	Low	Low	High
Zn in crude oils	Low	Intermediate	High
Hg in crude oils	Low	Low	High

It is tempting to speculate that the Zn in crude oils (and probably also Hg; see discussion in Hitchon and Filby, 1983, pp. 16-18) as well as Zn in formation waters, have originated from shales which yielded the crude oil hydrocarbons through the specific action of hot, chloride-rich formation waters, with the consequent depletion of those shales in Zn (and probably also Hg). The statistics on Cd in shales suggests little difference between the three groups, however.

A stepwise discriminant analysis of the shale composition data was carried out using logarithmic transforms where appropriate and the Wilk's lambda criterion for order of entry of the variables. Only Zn and Ca were significant discriminators, in order of their entry. Slightly more than 69% of the grouped samples were correctly classified. This result contrasts with that of Hitchon and Filby (1984) for grouped crude oils, where S, Co, Se, V and Br were the significant discriminators and 89.1% of the crude oils were correctly classified on the basis of their trace element content. The significance of these different results on shales and crude oils remains enigmatic.

## Vanadium

## General Statistics

V (ppm or mg/l)	Number of samples reporting element	Minimum	Median	Average	Maximum
GROUP 1					
Shales (n=7)	7	46	87	86	127
Crude oils (n=20)	20	0.003	0.21	0.834	6,553
GROUP 2				-	
Shales (n=12)	12	55	75	116	421*
Crude oils (n=25)	23	0.068	20.88	28.1	138.8
GROUP 3				, <u> </u>	·
Shales (n=7)	7	35	71	68	101
Crude oils (n=33)	32	0.014	1.0	6.28	52.67

<sup>\*</sup>without this extreme value (>3 $\sigma$  from the mean for all shales) the median, average and maximum values are 74, 88 and 165 ppm, respectively.

# Geochemistry

The relation of V in Alberta shales and crude oils was discussed by Hitchon and Filby (1983, pp. 21-25), and will not be repeated here. Suffice to note there is some evidence to support the suggestion of a relation between the content of V in crude oils and the content of V in the source rocks.

## Chromium

## General Statistics

Cr (ppm or mg/l)	Number of samples reporting element	Minimum	Median	Average	Maximum
GROUP 1					
Shales (n=7)	7	113	150	152	200
Crude oils (n=20)	7	0.00371	0.00780	0.00839	0.01551
Formation waters (n=15)	0	-	15	1/2	-
GROUP 2					
Shales (n=12)	12	116	160	162	228
Crude oils (n=25)	14	0.00466	0.245	0.0416	0.1058
Formation waters (n=25)	1	-	0.012	9	-
GROUP 3					
Shales (n=7)	7	44	101	114	221
Crude oils (n=33)	16	0.00363	0.0148	0.0326	0.2058
Formation waters (n=34)	3	0.016	0.016	0.022	0.034

# Geochemistry

Little can be said about Cr in shales, crude oils and formation waters except that (1) there are no obvious trends among the groups, (2) Cr is very low in formation waters, (3) Cr is present in small amounts in just less than half the crude oils analysed, and (4) Cr occurs in fairly uniform amounts in most shales.

## Manganese

## General Statistics

Mn (ppm or mg/l)	Number of samples reporting element	Minimum	Median	Average	Maximum
GROUP 1					
Shales (n=7)	7	74	221	284	740
Crude oils (n=20)	20	0.001	0.0506	0.0549	0.1192
Formation waters (n=15)	14	0.04	0.11	0.37	2.50
GROUP 2		,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,			
Shales (n=12)	12	104	241	310	921
Crude oils (n=25)	19	0.0017	0.0498	0.0471	0.07928
Formation waters (n=25)	22	0.09	0.33	0.53	3.18
GROUP 3					
Shales (n=7)	7	68	130	226	634
Crude oils (n=33)	31	0.01793	0.0492	0.0526	0.1766
Formation waters (n=34)	32	0.05	0.60	1.01	11.0

# Geochemistry

Manganese was present in nearly all samples analysed. It formed a separate factor in the formation waters studied by Hitchon et al. (1971), and a separate factor in the crude oils studied by Hitchon et al. (1975). The above data on Mn in shales suggest fairly uniform contents in all shales without obvious differences among the three groups.

# Iron, Cobait and Nickel

## General Statistics

Fe, Co, Ni (ppm or mg/l)	Number of samples reporting element	Minimum	Median	Average	Maximum
GROUP 1					
Shales (n=7)	Fe 7	20100	36000	35400	45600
	Co 6	5	19	17	28
	Ni 7	66	170	147	200
Crude oils (n=20)	Fe 10	0.071	0.466	0.545	1.671
	Co 18	0.000261	0.00271	0.00803	0.02907
	Ni 15	0.022	0.520	0.939	4.03
Formation waters (n=15)	Fe 15	0.46	2.7	5.8	27.2
	Co 1		0.009		
·	Ni 2	0.047	0.057	0.057	0.066
GROUP 2					
Shales (n=12)	Fe 12	14000	32700	35200	81100*
	Co 11	2	12	17	42
	Ni 12	78	133	133	195
Crude oils (n=25)	Fe 12	0.286	0.587	0.94	2.19
	Co 25	0.000397	0.004484	0.0102	0.0622
	Ni 22	0.136	8.53	11.5	57.48
Formation waters (n=25)	Fe 25	0.05	0.94	8.6	54
	Co 3	0.005	0.008	0.014	0.029
	Ni 7	0.028	0.071	0.063	0.110
GROUP 3	"				
Shales (n=7)	Fe 7	2730	25200	26700	49000
	Co 6	2	16	22	48
	Ni 7	48	150	167	350**
Crude oils (n=33)	Fe 18	0.216	0.566	2.04	19.77
	Co 32	0.000516	0.00866	0.0227	0.2026
	Ni 24	0.085	3.51	7.96	70.1
Formation waters (n=34)	Fe 32	0.01	0.54	51	42
	Co 7	0.005	0.012	0.011	0.015
	Ni 5	0.026	0.042	0.038	0.046

<sup>\*</sup> without this extreme value (>3σ from the mean for all shales) the median, average and maximum values are 32,000, 31,000 and 44,900 ppm, respectively.
\*\* without this extreme value (>3σ from the mean for all shales) the median, average and maximum values are 140, 136 and 260 ppm, respectively.

## Geochemistry

Iron, Co and Ni are present in nearly all shales, with Fe >> Ni > Co, with no significant differences between the groups except for slightly lower Fe in group 3 samples. Unlike the shales, the proportions in crude oils are Ni > Fe >> Co, with group 2 crude oils (which are heavy oils subjected to biodegradation and water washing) having enhanced contents of Ni. Although Fe is present in most formation waters, Co and Ni are seldom much above detection limits. Generally, there appears to be no relation among shales, crude oils and formation waters with respect to Fe, Co and Ni and the only differences among the groups related to lower Fe in group 3 shales and higher Ni in group 2 crude oils.

## Copper and Silver

#### General Statistics

Cu, Ag (ppm or mg/l)	Number of samples reporting element	Minimum	Median	Average	Maximum
GROUP 1					
Shales (n=7)	Cu 7	9	39	266	1600**
	Ag 3	1.2	1.3	16	44
Formation waters (n=15)	Cu 15	0.04	0.08	0.10	0.26
GROUP 2					
Shales (n=12)	Cu 12	21	31	34	70
	Ag 5	0.5	1.4	1.5	3.7
Formation waters (n=25)	Cu 25	0.05	0.11	0.14	0.44
GROUP 3					
Shales (n=7)	Cu 7	7	17	17	29
	Ag 5	0.5	2.4	4.8	16
Formation waters (n=34)	Cu 34	0.02	0.09	0.12	0.49

<sup>\*</sup> without this extreme value (>3σ from the mean for all shales) the median, average and maximum values are 35, 44 and 94, respectively.

# Geochemistry

Copper is present in all shales and formation waters, and Ag in half the shales. Ignoring the extreme content of Cu (1600 ppm) in sample K-1, the general content of Cu decreases with depth from group 1 to group 3. In contrast, there is little difference in the distribution of Cu in formation waters among the groups. The Ag content of K-1 is 44 ppm and has biased the statistics for Ag in group 1. Generally, however, there appears to be little difference among the groups, bearing in mind Ag was below detection in half the shales.

## Molybdenum

## General Statistics

Mo (ppm or mg/l)	Number of samples reporting element	Minimum	Median	Average	Maximum
GROUP 1			*		
Shales (n=7)	7	7	21	21	32
GROUP 2					
Shales (n=12)	11	3	11	12	24
GROUP 3					
Shales (n=7)	6	10	16	21	54*

<sup>\*</sup>without this extreme value (>3σ from the mean for all shales) the median, average and maximum values are 12.5, 12.2 and 20 ppm, respectively.

## Geochemistry

Molybdenum is found in nearly all shales. Even with the exception of sample K-10 (54 ppm Mo), there is no indication of significant differences among the groups.

General Statistics

Lead

Pb (ppm or mg/l)	Number of samples reporting element	% samples with Pb below detection	Minimum	Median	Average	Maximum
GROUP 1		=				
Shales (n=7)	7	0	5	10	17	55
GROUP 2						
Shales (n=12)	8	33	5	5	7	10
GROUP 3						
Shales (n=7)	0	100	-	-	-	-

# Geochemistry

Lead presents possibly the clearest evidence of trace element differences between the shale groups, made more obvious in the above table by the addition of the column of percentage of samples with Pb below detection (~5 ppm). The most obvious reason for this difference is that Pb occurs dominantly as chloride complexes in saline formation waters and so can easily be mobilized from the shales by the deeper, more saline formation waters.

## GEOCHEMISTRY OF KEROGEN

#### **ELEMENTAL COMPOSITION**

The raw elemental composition was adjusted for pyrite-S (% Fe x 1.1482) and ash, and normalized (Sum CHONS = 100). Orr (1986) showed that this method of adjusting for pyrite-S is in reasonable agreement with pyrite-S recovered by HNO<sub>3</sub> leaching.

When plotted (Figure 3B) on a van Krevelen diagram (atomic H/C against atomic O/C) the majority of samples are classed as Type III kerogen with three classed as Type II. Most fall in the zones of catagenesis and cracking (Figure 3A), i.e. the principal zone of oil formation and the zone of wet gas, respectively. Figure 3C shows the general environment of deposition of the shale from which the kerogen was extracted together with the estimated maximum temperature from Table 4. As expected, increased temperature results first in a rapid decrease in atomic O/C ratio (well seen in the non-marine Type III data), then a gradual decrease in the atomic H/C ratio for Type III kerogen.

On a ternary diagram of the heteroatoms (NSO = 100 wt.%) most data plot in the O-rich portion of the diagram (Figure 4). According to the proposed criteria of Orr (1986), all the Alberta Basin kerogens except BH-431G (atomic S/C 0.0234) would be classed as low-S types (atomic S/C < 0.02). Maximum temperature data from Table 4 are also given in Figure 4, and there is a tendency for a relative decrease in O in kerogen with increasing temperature. The solid tie line and dashed tie line join up data from the Viking Formation and non-oilsand Mannville Group, respectively. These trends suggest a tendency for a loss of both O and S with increased temperature, and therefore a relative increase in N as a residual effect. Plots (not shown) of both atomic H/C vs atomic N/C and atomic H/C vs atomic S/C are relatively 'flat'.

## **ESR PROPERTIES**

Three properties of the ESR signal were measured and are plotted against estimated maximum temperature in Figure 5A (broad band spins per gram), Figure 5B (g-value) and Figure 5C (broad band width). As expected, increased temperature results in increased spin concentrations, lower g-values, and wider lines. There seem to be distinct and separate trends for the Viking Formation and non-oilsand Mannville Group kerogens, with higher atomic H/C ratios (Figure 3B), higher relative amounts of O and lower relative amounts of S (Figure 4), and lower spin concentrations and g-values in the Viking Formation samples. The data are too few and the trends too tentative to do more than suggest the above relations.

## <sup>13</sup>C NMR PROPERTIES

Figure 6 shows three typical <sup>13</sup>C NMR spectra from kerogen samples analyzed in this study. The most useful parameter is the aliphatic-carbon/aromatic-carbon ratio, which

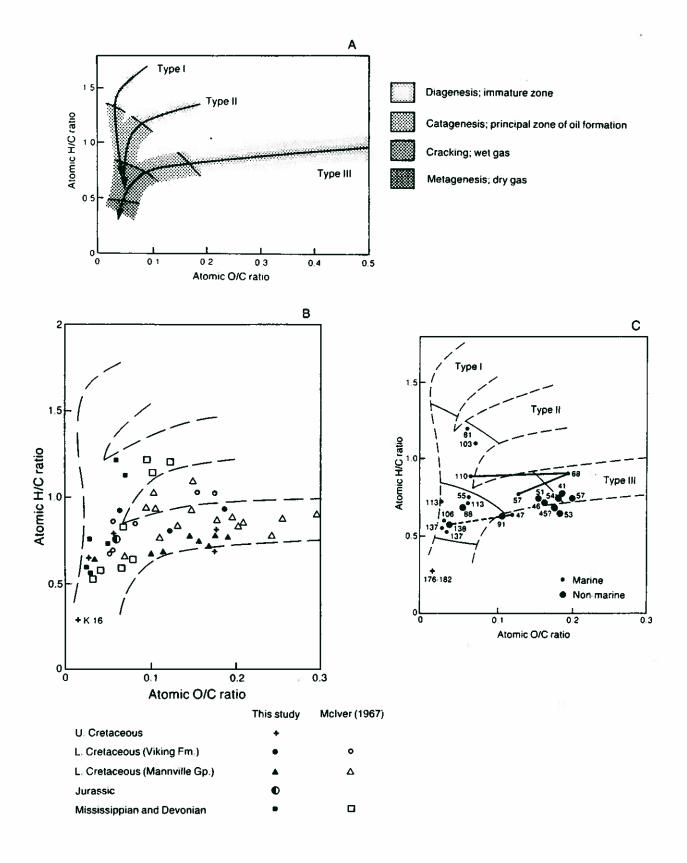


Figure 3. A. van Krevelan diagram (atomic H/C vs atomic O/C) showing maturation regions; B. van Krevelen diagram for kerogens from the Alberta Basin. The cross (+) is sample K-16 from the Middle Devonian Nahanni Formation of northeast British Columbia (4125 m, R<sub>o</sub> ~ 2.5, estimated present reservoir temperature 180°C). C. van Krevelan diagram for kerogens from the Alberta Basin showing the general environment of deposition of the shales from which the kerogen was extracted, and the estimated maximum temperatures (°C). Solid tie line - Viking Formation; dashed tie line - non-oilsands Mannville Group.

Table 4. Estimated maximum burial depth and temperature.

Sample number	Present depth (m) [Appendix A]	Eroded thickness (m) [Hitchon, 1984 Fig. 23]	Paleogeothermal gradient (°C/km) [Hitchon, 1984 Fig. 24]	Maximum burial depth (km)	Maximum temperature (°C)
RCA 65-1	80	2050	25	2.1	53
RCA 67-1	140	1400	27	1.5	41
K-1	2145	2500	23	4.6	106
W-28	440	*2000	*23	2.4	55
K-2	620	1500	27	2.1	57
K-3	1040	1600	26	2.6	68
K-4	2190	2400	24	4.6	110
BH-431A	225	*1500	*27	1.7	46
BH-431G	425	*1500	*27	1. <del>9</del>	51
BH-427A	370	1450	26	1.8	47
BH-439A	440	*1600	*27	2.0	54
BH-433A	455	*1600	*27	2.1	57
BH-435F	300	?	?	-	*45
K-5	1485	2000	26	3.5	91
K-6	3375	2600	23	6.0	138
K-7	2535	2400	23	4.9	113
K-8	1980	*2100	*25	4.1	103
HE 32/35	3120	2600	24	5.7	137
HE 36	3120	2600	24	5.7	137
K-9	955	1500	26	2.5	65
K-10	1895	2050	25	3.9	98
K-11	1565	1450	27	3.0	81
K-12	2805	1700	25	4.5	113
K-13	1620	1750	26	3.4	88
K-15	2655	2000	25	4.7	118

<sup>\* =</sup> estimated

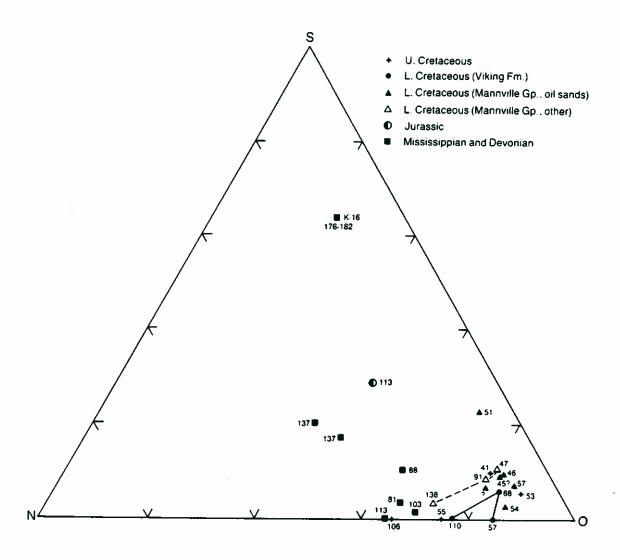
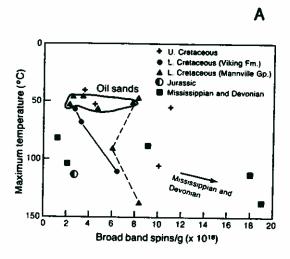


Figure 4. Ternary diagram of heteroatoms (NSO=100 wt.%) in kerogens from the Alberta Basin, showing stratigraphic unit and estimated maximum temperature (°C). For details of sample K-16 see caption to Figure 3. Solid tie line - Viking Formation; dashed tie line - non-oilsands Mannville Group.



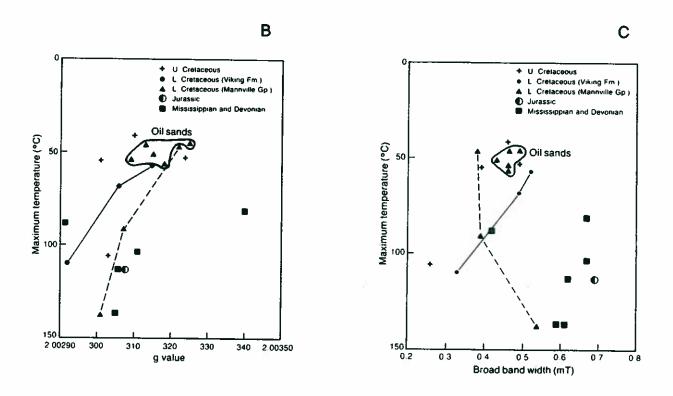


Figure 5. Relation of: A. ESR broad band spin concentration (spins/gx10<sup>18</sup>); B. ESR g-value; and C. ESR broad band width (mT) to estimated maximum temperature (°C) in kerogens from the Alberta Basin. Solid tie line - Viking Formation; dashed tie line - non-oilsands Mannville Group.

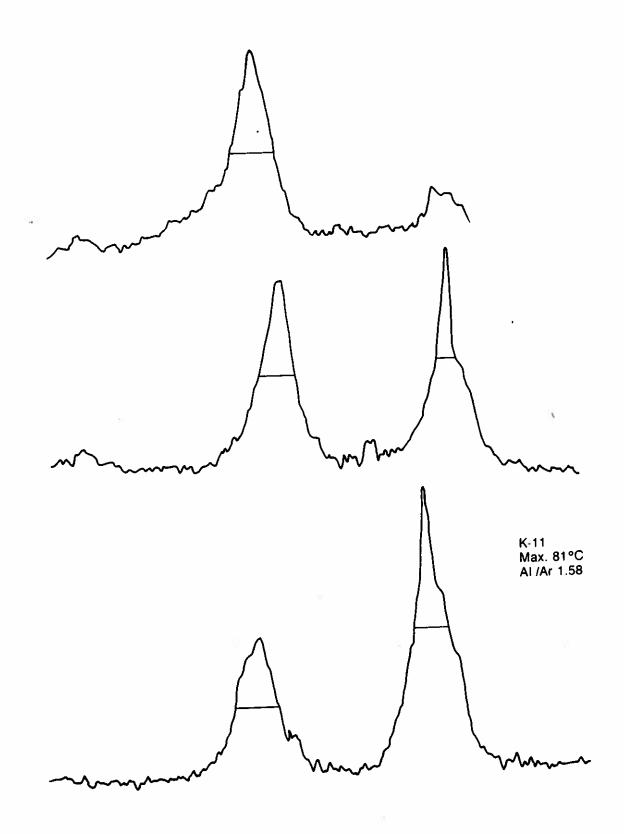


Figure 6. Typical <sup>13</sup>C NMR spectra from kerogen samples from the Alberta Basin.

reflects both the original chemical composition of the kerogen and a measure of the maturation to which it has been subjected. This ratio is reported in Table 5 for the samples studied. Figure 7 is a plot of the aliphatic/aromatic ratio against estimated maximum temperature. For the three samples from the Viking Formation increasing temperature results in generally decreased aliphatic/aromatic carbon ratio; this trend is not obvious for the other samples, and at any specific temperature there is a wide range of ratios.

#### **ACKNOWLEDGEMENTS**

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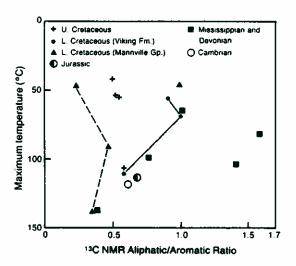


Figure 7. Relation of aliphatic-carbon/aromatic-carbon ratio and estimated maximum temperature (°C) in kerogens from the Alberta Basin. Solid tie line - Viking Formation; dashed tie line - non-oilsands Mannville Group.

Table 5. <sup>13</sup>C NMR properties of kerogen extracted from Alberta Basin shales; for complete details of samples see Appendix A.

Sample number	<sup>13</sup> C NMR properties			
	width (mT)	aliphatic/aromatic carbon ratio (corrected for spinning side bands)		
RCA 65-1	0.49	0.52		
RCA 67-1	0.46	0.50		
K-1	0.26	0.58		
W-28	0.39	0.54		
K-2	0.52	0.92		
K-3	0.49	1.00		
K-4	0.33	0.58		
BH-431A	0.49	•		
BH-431G	0.43			
BH-427A	0.38	0.23		
BH-439A	0.46			
BH-433A	0.46			
BH-435F	0.46	0.99		
K-5	0.39	0.47		
K-6	0.54	0.35		
K-7	0.69	0.67		
RL 2/4	-	-		
K-8	0.67	1.41		
HE 32/35	0.61	0.38		
HE 36	0.59	S .		
K-9	-	1.0		
K-10	-	0.76		
K-11	0.67	1.58		
K-12	0.62	-		
K-13	0.42			
K-14	-	<u>-</u>		
K-16	-			
K-15	-	0.61		

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# APPENDIX A

Geochemistry of Alberta shales and associated kerogen

Data sheets

#### GEOCHEMISTRY OF ALBERTA SHALES AND ASSOCIATED KEROGEN: DATA SHEET 1

Sample number: RCA 65-1

Stratigraphic unit: U.Cretaceous, Post-Colorado Supergroup,

Edmonton Fm.

Well name: RCA Corehole 65-1 (Wizard Lake) 4-8-48-27-W4M Depth (m): 79.2-80.5 claystone, silty, grey, shaley partings fresh to brackish water Location: Description:

Depositional environment:

MAJOR OXIDES (%)		ASSOCIATED KEROGEN		
SiO <sub>2</sub>	71	Normalized composition (S	Sum CHONS = 100)	
TiO <sub>2</sub>	0.7	Adjusted for pyrite sulfur (%Fe x 1.1482)		
Al <sub>2</sub> O <sub>3</sub>	17			
Σ Fe as Fe <sub>z</sub> O <sub>3</sub>	5.7			
MgO	2.0	C	75.30	
CaO	1.2	Н	4.33	
Na <sub>2</sub> O	0.7	0	17.77	
	Σ 98.3	N	1.45	
TRACE ELEMENTS	(ppm)	\$	1.15	
Ag	•			
Cd	14	Atomic H/C	0.685	
Co	26	O/C	0.1771	
Cr	164	N/C	0.0164	
Cu	70	S/C	0.0057	
Mn	740			
Мо	21	ESR properties (in vacuo,	corrected for pyrite)	
Ni	90			
Pb	10	spins/g Narrow	•	
Sr	170	Broad	4.5x10 <sup>18</sup>	
V	87	g-value Narrow	2.00300	
Zn	161	Broad	2.00324	
		width (mT) Narrow	0.13	
<u>ORGANICS</u>		Broad	0.49	
Organic C (%)	1.116			
Extractable		SULFUR ISOTOPES (824S	% CDM)	
organics (ppm)	80.5			
R <sub>e</sub> (average)	0.70	Kerogen	-8.8	
		Extractable organics	-0.4	
* = Below detection		Pyrite	-	

#### **COMMENTS**

- = Not determined

For a description of the cored section see Snead (1969, pp. 74-80) and Carrigy (1971, pp. 97-100). Sample probably best described as a siltstone.

## GEOCHEMISTRY OF ALBERTA SHALES AND ASSOCIATED KEROGEN: DATA SHEET 2

Sample number: Stratigraphic unit: Well name: Location: Description: Depositional environal environa		Bearpaw Fm. RCA Corehole No. 13-34-37-13-W4N medium to dark g	M Depth (m) 137.8-141.  prey, bentonitic shale quiet shallow water  ASSOCIATED KEROG	<u>EN</u> n (SUM CHONS = 100)
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	5.8			
MgO	1.7		С	73.31
CaO	0.9		Н	4.92
Na <sub>2</sub> O	0.7		0	17.41
	Σ 100.7		N	2.21
TRACE ELEMENT	S (ppm)		S	2.15
Ag	•			
Cd	13		Atomic H/C	0.800
Co	26		O/C	0.1783
Cr	149		N/C	0.0257
Cu	39		S/C	0.0110
Mn	301			
Mo	32		ESR properties (in vacu	io, corrected for pyrite)
Ni	120			
Pb	10		spins/g Narrow	1.1x10 <sup>16</sup>
Sr	170		Broad	3.6x10 <sup>18</sup>
V	98		g-value Narrow	2.00280
Zn	154		Broad	2.00310
			width (mT) Narrow	-
<u>ORGANICS</u>			Broad	0.46
Organic C (%)	1.088			
Extractable			SULFUR ISOTOPES (8	<sup>34</sup> S ‰ CDM)
organics (ppm)	980			
R <sub>o</sub> (average)	0.68		Kerogen	-12.1
			Extractable organics	-3.8

## COMMENTS

\* = Below detection

- = Not determined

For a description of the well see Given and Wall (1971, p. 507; well log Fig. 2; microfaunal-microfloral associations Fig. 6). Sample came from the middle of the Lower Shale Unit.

Pyrite

-7.9

Sample number: Stratigraphic unit: Well name: Location: Description: Depositional envir	onment:	Canadian Superi	.Colorado Gp., Cardium F or Los Nietos O'Chiese 9- l Depth (m): 2137.6-2150	-11
MAJOR OXIDES	<u>(%)</u>		ASSOCIATED KEROGE	<u>N</u>
SiO <sub>2</sub>	60		Normalized composition	(SUM CHONS = 100)
TiO <sub>2</sub>	0.6		Adjusted for pyrite sulfur	(% Fe x 1.1482)
Al <sub>2</sub> O <sub>3</sub>	10			
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	4.3			
MgO	1.2		С	89.66
CaO	0.4		Н	4.75
Na <sub>2</sub> O	0.2		0	3.69
	Σ 76.7		N	1.90
TRACE ELEMENT	rs (ppm)		S	0.00
Ag	44			
Cd	13		Atomic H/C	0.631
Co	5		O/C	0.0309
Cr	130		N/C	0.0181
Cu	1600		S/C	- £
Mn	164			
Мо	20	*0	ESR properties (in vacuo	o, corrected for pyrite)
Ni	180			
РЬ	55		spins/g Narrow	-
Sr	64		Broad	1.0x10 <sup>19</sup>
V	78		g-value Narrow	2.00268
Zn	943		Broad	2.00303
			width (mT) Narrow	-
ORGANICS			Broad	0.26
Organic C (%)	0.816			
Extractable			SULFUR ISOTOPES (83	'S ‰ CDM)
organics (ppm)	1900			
R, (average)	1.27		Kerogen	-
			Extractable organics	-5.1
* = Below detection	n		Pyrite	-
- = Not determined	I			

# COMMENTS

Core 2, Box 4 of 10.

Sample number:

W-28

Stratigraphic unit:

U.Cretaceous, U.Colorado Gp., Kaskapau Fm.

Doe Creek Ss.

Well name:

Pacific Inland PCoupeS 9-5DC-78-12

Location: Description: 9-5-78-12-W6M

Depth (m): 440.4-443.5

Depositional environment:

marine

Debosingual étim	Omneric.	maine		
MAJOR OXIDES	<u>(%)</u>			ASSOCIATED KEROGEN
SiO <sub>2</sub>	60		Normalized composition	n (SUM CHONS=100)
TiO <sub>2</sub>	0.9		Adjusted for pyrite sulfu	иг (% Fe x 1.1482)
Al <sub>2</sub> O <sub>3</sub>	18			
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	5.1			
MgO	1.5		C	85.46
CaO	0.4		H	5.52
Na <sub>2</sub> O	0.4		0	6.79
	Σ 86.3		N	2.23
TRACE ELEMEN	TS (ppm)		s	0.00
Ag	1.2			
Cd	11		Atomic H/C	0.770
Со	11		O/C	0.0596
Cr	200		N/C	0.0223
Cu	94		S/C	•
Mn	192			
Мо	7		ESR properties (in vacu	uo, corrected for pyrite)
Ni	66			
Pb	15		spins/g Narrow	1.5x10 <sup>17</sup>
Sr	113		Broad	1.1x10 <sup>19</sup>
V	127		g-value Narrow	2.00279
Zn	206		Broad	2.00301
			width (mT) Narrow	-
<u>ORGANICS</u>			Broad	0.39
Organic C (%)	1.400			
Extractable			SULFUR ISOTOPES (8	34S % CDM)
organics (ppm)	2780			
R <sub>o</sub> (average)	-		Kerogen	+19.7
			Extractable organics	+16.6
* = Below detection	on		Pyrite	+29.0

# COMMENTS

- = Not determined

 $\delta^{18}O$  (‰ SMOW) = +18.5 on carbonate cement from adjacent sandstone (Hitchon and Friedman, 1969, p. 1344, sample No. 28).

Sample number:

K-2

Stratigraphic unit:

L.Cretaceous, L.Colorado Gp., Viking Fm.

Well name: Location:

Imperial Pinedale No. 1

7-28-54-16-W4M

Depth (m): 602.0-604.7 633.4-635.5

Description:

medium grey shale

Depositional environment:

marine

MAJOR OXIDES (	<u>%)</u>	ASSOCIATED KEROC	<u>GEN</u>	
SiO <sub>2</sub>	68	Normalized composition	on (SUM CHONS = 100)	
TiO <sub>2</sub>	0.7	Adjusted for pyrite sulf	fur (% Fe x 1.1482)	
Al <sub>2</sub> O <sub>3</sub>	17			
Σ Fe as Fe <sub>2</sub> O <sub>3</sub>	6.5			
MgO	2.0	С	79.29	
CaO	0.4	Н	5.38	
Na₂O	0.5	0	13.03	
	Σ 95.1	N	2.30	
TRACE ELEMENT	S (ppm)	S	0.00	
Ag	•			
Cd	24	Atomic H/C	0.808	
Co	5	O/C	0.1234	
Cr	150	N/C	0.0247	
Cu	23	S/C	-	
Mn	221			
Мо	26	ESR properties (in vac	uo, corrected for pyrite)	
Ni	200			
Pb	15	spins/g Narrow	-	
Sr	180	Broad	2.8x1018	
V	66	g-value Narrow	2.00289	
Zn	145	Broad	2.00315	
		width (mT) Narrow	-	
ORGANICS		Broad	0.52	
Organic C (%)	1.176			
Extractable		SULFUR ISOTOPES (	SULFUR ISOTOPES (834S % CDM)	
organics (ppm)	1030			
R <sub>e</sub> (average)	0.79	Kerogen	-11.1	
		Extractable organics	-4.8	
* = Below detection	1	Pyrite	-14.8	

## **COMMENTS**

- = Not determined

Sample number: Stratigraphic unit: Well name:

K-3

L.Cretaceous, L.Colorado Gp., Viking Fm.

Depth (m): 1038.5-1040.0

Imp Home Duhamel 4-8-46-21

Location: Description: 4-8-46-21-W4M medium grey shale

Depositional environment:

marine

Depositional envi	oment.	manne		
MAJOR OXIDES	<u>(%)</u>		ASSOCIATED KEROG	<u>EN</u>
SiO <sub>2</sub>	71		Normalized composition	n (SUM CHONS = 100)
TiO₂	0.8		Adjusted for pyrite sulfu	ır (% Fe x 1.1482)
Al <sub>2</sub> O <sub>3</sub>	15			
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	5.1			
MgO	1.6		C	72.58
CaO	0.3		н	5.71
Na₂O	0.6		0	18.10
	Σ 94.4		N	2.34
TRACE ELEMEN	TS (ppm)	•	S	1.27
Ag	•			
Cd	26		Atomic H/C	0.937
Co	28		O/C	0.1872
Cr	160		N/C	0.0275
Cu	30		S/C	0.0066
Mn	297			
Мо	28		ESR properties (in vacu	o, corrected for pyrite)
Ni	200			
Pb	10		spins/g Narrow	-
Sr	183		Broad	3.3x10 <sup>18</sup>
V	102		g-value Narrow	2.00277
Zn =	180		Broad	2.00306
			width (mT) Narrow	•
<b>ORGANICS</b>			Broad	0.49
Organic C (%)	1.460			
Extractable			SULFUR ISOTOPES (8	<sup>34</sup> S ‰ CDM)
organics (ppm)	815			
R <sub>e</sub> (average)	0.70		Kerogen	-7.2
			Extractable organics	-1.6
* = Below detection	n		Pyrite	+2.3

## COMMENTS

- = Not determined

K-4

Sample number: Stratigraphic unit: Well name: L.Cretaceous, L.Colorado Gp., Viking Fm.
Oilwell Operators et al. Alhambra 16-11
16-11-38-5-W5M
De

Location: Depth (m): 2188.5-2190.0

Description: medium grey shale

Depositional environment: marine

MAJOR OXIDES (%)		ASSOCIATED KEROGEN	
SiO <sub>2</sub>	56	Normalized composition (	SUM CHONS = 100)
TiO <sub>2</sub>	0.5	Adjusted for pyrite sulfur (	(% Fe x 1.1482)
Al <sub>2</sub> O <sub>3</sub>	9		
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	2.9		
MgO	1.2	С	84.33
CaO	0.5	н	6.52
Na₂O	0.1	0	7.10
	Σ 70.2	N	2.05
TRACE ELEMENTS	G (ppm)	S	0.00
Ag	1.3		
Cd	6.4	Atomic H/C	0.921
Co	•	O/C	0.0632
Cr	113	N/C	0.0207
Cu	9	S/C	40
Mn	74		
Мо	10	ESR properties (in vacuo, corrected for pyrite)	
Ni	170		
Pb	5	spins/g Narrow	-
Sr	62	Broad	6.4x10 <sup>18</sup>
V	46	g-value Narrow	2.00287
Zn	99	Broad	2.00292
		width (mT) Narrow	•
<u>ORGANICS</u>		Broad	0.33
Organic C (%)	1.092		
Extractable		SULFUR ISOTOPES (824S % CDM)	
organics (ppm)	2090		
R <sub>e</sub> (average)	0.88	Kerogen	+6.1
		Extractable organics	+12.1
* = Below detection		Pyrite	+2.8
- = Not determined			

## COMMENTS

BH-431A

Stratigraphic unit Well name: Location: Description: Depositional envi		- L.Cretaceous, Gi BA Wabasca 10- 10-25-81-22-W4I medium to dark of non-marine	25-81-22	Depth (m): 222.5-226.8
MAJOR OXIDES	<u>(%)</u>		ASSOCIATED KEROC	<u>GEN</u>
SiO <sub>2</sub>	71		Normalized composition	on (SUM CHONS = 100)
TiO <sub>2</sub>	8.0		Adjusted for pyrite sulf	ur (% Fe x 1.1482)
Al <sub>2</sub> O <sub>3</sub>	18			
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	5.4			
MgO	2.2		C	75.53
CaO	0.3		Н	4.75
Na <sub>2</sub> O	0.6		0	15.85
	Σ 98.3		N	2.59
TRACE ELEMEN	ITS (ppm)		s	1.28
Ag	•			
Cd	14		Atomic H/C	0.749
Co	2		O/C	0.1575
Cr	140		N/C	0.0292
Cu	21		S/C	0.0063
Mn	297			
Мо	7		ESR properties (in vac	uo, corrected for pyrite)
Ni	101			
Pb	10		spins/g Narrow	-
Sr	207		Broad	2.6x10 <sup>18</sup>
V	59		g-value Narrow	
Zn	150		Broad	2.00313
			width (mT) Narrow	¥10
ORGANICS			Broad	0.49
Organic C (%)	0.900			
Extractable		SULFUR ISOTOPES (	6 <sup>24</sup> S ‰ CDM)	
organics (ppm)	555			
R <sub>o</sub> (average)	-		Kerogen	-16.5
			Extractable organics	-
* = Below detection	on		Pyrite	-28.2
- = Not determine	d			

## **COMMENTS**

Sample number:

Wabasca (Grand Rapids A) oil sand deposit; composite of 13 samples over interval 222.5-226.8 m.

Sample number:

BH-431G

Stratigraphic unit:

L.Cretaceous, Clearwater Fm., Wabiskaw Mbr.

Depth (m): 425.5-427.9

Well name:

BA Wabasca 10-25-81-22

Location:

10-25-81-22-W4M

Description:
Depositional environment:

carbonaceous shale above coal

non-marine

		11011-11102-1110		
MAJOR OXIDES (%)			ASSOCIATED KEROGEN	
SiO <sub>2</sub>	45		Normalized composition	(SUM CHONS = 100)
TiO <sub>2</sub>	0.6		Adjusted for pyrite sulfu	r (% Fe x 1.1482)
Al <sub>2</sub> O <sub>3</sub>	15			
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	4.2			
MgO	0.9		С	74.67
CaO	0.6		Н	4.85
Na₂O	0.1		0	14.58
	Σ 66.4		N	1.24
TRACE ELEMENT	TS (ppm)		s	4.66
Ag	•			
Cd	10		Atomic H/C	0.774
Co	8		O/C	0.1466
Cr	158		N/C	0.0142
Cu	29		S/C	0.0234
Mn	231			
Мо	6		ESR properties (in vacu	o, corrected for pyrite)
Ni	78			
Pb	•		spins/g Narrow	7.8x10 <sup>16</sup>
Sr	95		Broad	7.9x10 <sup>1a</sup>
V	59		g-value Narrow	2.00295
Zn	127		Broad	2.00315
			width (mT) Narrow	•
ORGANICS			Broad	0.43
Organic C (%)	2.020			
Extractable		SULFUR ISOTOPES (834S % CDM)		
organics (ppm)	2070			
R <sub>e</sub> (average)	0.67		Kerogen	+18.5
			Extractable organics	+24.1
* = Below detection	n		Pyrite	+21.7
<b>**-*</b> - <b>!</b> -* *				

## **COMMENTS**

- = Not determined

Wabasca oil sand deposit; composite of 10 samples over sampled interval.

Depth (m): 365.2-379.5

79.83

4.59

12.48

1.54

1.56

0.685

0.1173

0.0165

0.0073

8.2x10<sup>18</sup>

2.00284

2.00322

0.38

Sample number: BH-427A Stratigraphic unit: L.Cretaceous, Clearwater Fm. Well name: Union Red Earth 10-14-88-9 Location: 10-14-88-9-W5M Description: medium grey shale with pyrite nodules Depositional environment: marine MAJOR OXIDES (%) ASSOCIATED KEROGEN SiO, 49 Normalized composition (SUM CHONS = 100) TiO, 0.6 Adjusted for pyrite sulfur (% Fe x 1.1482) Al<sub>2</sub>O<sub>3</sub> 13 ΣFe as Fe<sub>2</sub>O<sub>3</sub> 11.6 MgO 4.7 C CaO 4.4 Н Na<sub>2</sub>O 0 0.1 Σ 83.4 N TRACE ELEMENTS (ppm) S Ag 0.5 Cd 22 Atomic H/C Co 11 O/C Cr 228 N/C Cu 40 S/C Mn 921 Мо 21 ESR properties (in vacuo, corrected for pyrite) Ni 159 РЬ spins/g Narrow Sr 333 **Broad** ٧ 145 g-value Narrow Zn 261 **Broad** width (mT) Narrow **ORGANICS** Broad Organic C (%) 0.844 Extractable SULFUR ISOTOPES (834S % CDM) organics (ppm) 865 R<sub>a</sub> (average) Kerogen

# COMMENTS

\* = Below detection

- = Not determined

Loon River oil sand deposit.

Extractable organics

**Pyrite** 

Sample number: Stratigraphic unit:

BH-439A

Well name:

L.Cretaceous, U.Grand Rapids Fm. Pan Am A1 Michel Lake 10-25-58-5

Location:

10-25-58-5-W4M

Depth (m): 441.4-443.2

Description:

light to medium gray, fissile to blocky shale

Depositional environment:

non-marine

MAJOR OXIDES	<u>(%)</u>	ASSOCIATED KEROG	ASSOCIATED KEROGEN	
SiO <sub>2</sub>	62	Normalized composition	(SUM CHONS = 100)	
TiO <sub>2</sub>	0.8	Adjusted for pyrite sulfu	ır (% Fe x 1.1482)	
Al <sub>2</sub> O <sub>3</sub>	19			
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	4.6			
MgO	2.1	С	74.58	
CaO	0.5	н	4.88	
Na <sub>2</sub> O	0.4	0	17.62	
	Σ 89.4	N	2.37	
TRACE ELEMENT	rs (ppm)	S	0.55	
Ag	1.4			
Cd	13	Atomic H/C	0.780	
Co	11	O/C	0.1773	
Cr	217	N/C	0.0271	
Cu	32	s/c	0.0028	
Mn	423			
Мо	7	ESR properties (in vacu	o, corrected for pyrite)	
Ni	113			
Pb	10	spins/g Narrow	-	
Sr	121	Broad	2.3x10 <sup>18</sup>	
٧	63	g-value Narrow	-	
Zn	109	Broad	2.00309	
		width (mT) Narrow	-	
ORGANICS		Broad	0.46	
Organic C (%)	0.432			
Extractable		SULFUR ISOTOPES (8	SULFUR ISOTOPES (834S % CDM)	
organics (ppm)	895			
R <sub>o</sub> (average)	•	Kerogen	-	
		Extractable organics	-	
* = Below detection	= Below detection Pyrite -		-	
		•		

## COMMENTS

- = Not determined

Cold Lake oil sand deposit.

Sample number: Stratigraphic unit: Well name: Location: Description: Depositional environment:

BH-433A

L.Cretaceous, Mannville Gp. Imp 73-5 Leming OV 13-33-64-3

13-33-64-3-W4M

Depth (m): 454.5-458.3

dark grey, slightly silty shale

non-marine

MAJOR OXIDES (%)		ASSOCIATED KEROGE	ASSOCIATED KEROGEN	
SiO₂	45	Normalized composition	(SUM CHONS = 100)	
TiO <sub>2</sub>	0.6	Adjusted for pyrite sulfu	r (% Fe x 1.1482)	
Al <sub>2</sub> O <sub>3</sub>	13			
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	6.4			
MgO	1.7	C	73.41	
CaO	0.4	н	4.76	
Na₂O	0.7	0	18.64	
	Σ 67.8	N	1.74	
TRACE ELEMENT	'S (ppm)	s	1.45	
Ag	•			
Cd	13	Atomic H/C	0.773	
Co	12	O/C	0.1906	
Cr	172	N/C	0.0202	
Cu	32	S/C	0.0074	
Mn	250			
Мо	11	ESR properties (in vacu	o, corrected for pyrite)	
Ni	78			
Pb	•	spins/g Narrow	2.6x10 <sup>16</sup>	
Sr	202	Broad	4.7x10 <sup>16</sup>	
V	59	g-value Narrow	2.00285	
Zn	109	Broad	2.00318	
		width (mT) Narrow	-	
ORGANICS		Broad	0.46	
Organic C (%)	3.924			
Extractable		SULFUR ISOTOPES (δ3	SULFUR ISOTOPES (834S % CDM)	
organics (ppm)	9550			
R <sub>o</sub> (average)	0.61	Kerogen	-8.7	
		Extractable organics	+3.6	
* = Below detection		Pyrite	-16.0	

## **COMMENTS**

- = Not determined

Cold Lake oil sand deposit (Lower Grand Rapids Fm.).

Sample number:

BH 435-F

Stratigraphic unit:

 L.Cretaceous, McMurray Fm. Shell Ells OV 6-25-95-16

Depth (m): 301.1-301.4

Well name: Location:

6-25-95-16-W4M dark grey fissile shale

Description:
Depositional environment:

freshwater deltaic

Depositional envir	OINTIGHT.	Heshwaler Genale		
MAJOR OXIDES	<u>(%)</u>		ASSOCIATED KEROG	<u>EN</u>
SiO <sub>2</sub>	64		Normalized composition	(SUM CHONS = 100)
TiO₂	0.7		Adjusted for pyrite sulfu	r (% Fe x 1.1482)
Al <sub>2</sub> O <sub>3</sub>	22			
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	3.4			
MgO	1.0		С	74.82
CaO	0.2		Н	4.46
Na₂O	0.3		0	17.01
	Σ 91.6		N	1.79
TRACE ELEMEN	TS (ppm)		s	1.92
Ag	•			
Cd	10		Atomic H/C	0.710
Co	11		O/C	0.1707
Cr	161		N/C	0.0204
Cu	32		S/C	0.0096
Mn	104			
Мо	3		ESR properties (in vacu	o, corrected for pyrite)
Ni	95			
Pb	5		spins/g Narrow	-
Sr	113		Broad	4.0x10 <sup>18</sup>
V	76		g-value Narrow	2.00288
Zn	99		Broad	2.00325
			width (mT) Narrow	-
ORGANICS			Broad	0.46
Organic C (%)	4.912			
Extractable	Extractable		SULFUR ISOTOPES (δ34S % CDM)	
organics (ppm)	3810			
R <sub>o</sub> (average)	-		Kerogen	-
			Extractable organics	+20.7
* = Below detection	ก		Pyrite	+23.1

# COMMENTS

- = Not determined

Athabasca oil sand deposit

Sample number:

K-5

Stratigraphic unit:

L.Cretaceous, U.Mannville Gp. Anglo Canadian Wabamun No. 1

Well name: Location:

5-10-51-4-W5M

Depth (m): 1481.6-1484.7

Description:

Depositional environment:

fluviatile-lagoonal (Luscar facies; see Mellon

(1967)

MAJOR OXIDES (%)		ASSOCIATED KEROGEN	
SiO <sub>2</sub>	58	Normalized composition (S	SUM CHONS = 100)
TiO <sub>2</sub>	0.8	Adjusted for pyrite sulfur (	% Fe x 1.1482)
Al <sub>2</sub> O <sub>3</sub>	14		
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	4.3		
MgO	2.0	С	81.49
CaO	0.5	Н	4.59
Na₂O	0.6	0	11.10
	Σ 80.2	N	1.68
TRACE ELEMENT	(S (ppm)	S	1.14
Ag	•		
Cd	16	Atomic H/C	0.671
Co	14	O/C	0.1022
Cr	140	N/C	0.0176
Cu	70	S/C	0.0052
Mn	620		
Мо	20	ESR properties (in vacuo,	corrected for pyrite)
Ni	180		
Pb	5	spins/g Narrow	-
Sr	120	Broad	6.0x10 <sup>18</sup>
V	74	g-value Narrow	2.00284
Zn	166	Broad	2.00307
		width (mT) Narrow	-
ORGANICS		Broad	0.39
Organic C (%)	1,656		
Extractable		SULFUR ISOTOPES (δ <sup>24</sup> S ‰ CDM)	
organics (ppm)	3520		
R <sub>e</sub> (average)	0.50	Kerogen	•
		Extractable organics	+2.9
* = Below detection		Pyrite	-
- = Not determined	1		

## COMMENTS

For a description of the lithology, fossil content and facies of this well see Mellon (1967, Figs. 11 and 13).

Sample number:

K-6

Stratigraphic unit:

L.Cretaceous, L.Blairmore Gp. Triad BA Stolberg 6-10A 6-10-42-15-W5M

Well name:

Location:

Depth (m): 3374.4-3375.1

Description:

Depositional environment: fluviatile (Mountain Park facies); see Mellon (1967)

MAJOR OXIDES (%)		ASSOCIATED KEROGEN	
SiO <sub>2</sub>	64	Normalized composition (	SUM CHONS = 100)
TiO <sub>2</sub>	0.5	Adjusted for pyrite sulfur (	% Fe x 1.1482)
Al <sub>2</sub> O <sub>3</sub>	11		
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	4.8		
MgO	0.8	С	89.56
CaO	0.9	н	4.74
Na₂O	0.1	0	4.10
	Σ 82.1	N	1.40
TRACE ELEMENTS	(ppm)	S	0.20
Ag	•		
Cd	14	Atomic H/C	0.631
Co	42	O/C	0.0344
Cr	120	N/C	0.0133
Cu	29	S/C	0.0008
Mn	121		
Мо	•	ESR properties (in vacuo, corrected for pyrite)	
Ni	150		
Pb	5	spins/g Narrow	3.0x10 <sup>16</sup>
Sr	111	Broad	8.4x10 <sup>18</sup>
V	93	g-value Narrow	2.00272
Zn	124	Broad	2.00301
		width (mT) Narrow	0.11
ORGANICS		Broad	0.54
Organic C (%)	2.068		
Extractable		SULFUR ISOTOPES (834S % CDM)	
organics (ppm)	2020		
R <sub>o</sub> (average)	0.47	Kerogen	+9.5
		Extractable organics	+6.6
* = Below detection		Pyrite	+9.4
- = Not determined			

# COMMENTS

Sample number:

K-7

Stratigraphic unit:

L.Jurassic, Fernie Gp., Poker Chip Shale

Depth (m): 2535.0-2539.9

Well name:

Union Condor 10-32-39-6

Location: Description: 10-32-39-6-W5M medium grey, silty shale

Depositional environment:

marine

MAJOR OXIDES (%)		ASSOCIATED KEROGEN			
SiO <sub>2</sub>	64	Normalized composition (S	SUM CHONS = 100)		
TiO <sub>2</sub>	0.8	Adjusted for pyrite sulfur (% Fe x 1.1482)			
Al <sub>2</sub> O <sub>3</sub>	18				
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	4.8				
MgO	3.1	С	81.28		
CaO	2.8	Н	5.23		
Na₂O =	0.0	0	6.45		
	Σ 93.5	N	3.22		
TRACE ELEMENTS	S (ppm)	\$	3.82		
Ag	0.5				
Cd	13	Atomic H/C	0.767		
Co	42	O/C	0.0596		
Cr	165	N/C	0.0338		
Cu	29	S/C	0.0176		
Mn	408				
Мо	18	ESR properties (in vacuo,	corrected for pyrite)		
Ni	130				
Pb	5	spins/g Narrow	1.0x10 <sup>17</sup>		
Sr	198	Broad	2.7x10 <sup>18</sup>		
V	55	g-value Narrow	2.00277		
Zn	65	Broad	2.00307		
		width (mT) Narrow	0.03		
ORGANICS		Broad	0.69		
Organic C (%)	0.680				
Extractable		SULFUR ISOTOPES (δ34S	‰ CDM)		
organics (ppm)	1650				
R <sub>o</sub> (average)	•	Kerogen	-		
		Extractable organics	-		
* = Below detection		Pyrite	-		

## **COMMENTS**

- = Not determined

For details of the core from this well see Deere and Bayliss (1969)

Sample number:

RL 2/4

Stratigraphic unit:

Mississippian, Banff Fm.

Well name:

Imperial Rainbow Lake 16-18-107-6

Location:

16-18-107-6-W6M Depth (m): 542.7-546.9 medium dark grey shale (alternating with light grey

Description: medium dark grey shale (alternating w limestone with scattered brachiopods)

Depositional environment:

marine

Depositional env	ironment.	manne				
MAJOR OXIDES	<u> (%)</u>		ASSOCIATED KEROGEN			
SiO <sub>2</sub>	81(?)		Normalized composition	n (SUM CHONS = 100)		
TiO <sub>2</sub>	1.2		Adjusted for pyrite sulf	ur (% Fe x 1.1482)		
Al <sub>2</sub> O <sub>3</sub>	21					
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	5.0					
MgO	2.6		C	79.24		
CaO	2.2		н	5.23		
Na <sub>2</sub> O	0.4		0	13.69		
	Σ 113.4?		N	1.84		
TRACE ELEMEN	ITS (ppm)		S	0.00		
Ag	1.4					
Cd	15		Atomic H/C	0.786		
Co	17		O/C	0.1297		
Cr	200		N/C	0.0198		
Си	40		S/C	-		
Mn	110					
Мо	7		ESR properties (in vac	uo, corrected for pyrite)		
Ni	136					
Pb	•		spins/g Narrow	•		
Sr	224		Broad	7.7x10 <sup>16</sup>		
V	121		g-value Narrow	•		
Zn	120		Broad	2.00294		
			width (mT) Narrow	•		
<u>ORGANICS</u>			Broad	0.48		
Organic C (%)	0.604					
Extractable			SULFUR ISOTOPES (	δ <sup>34</sup> S ‰ CDM)		
organics (ppm)	1220					
R <sub>e</sub> (average)	-		Kerogen	-13.0		
			Extractable organics	•		
* = Below detecti	on		Pyrite	-13.7		
- = Not determine	ed					

# COMMENTS

Sum of CHONS of kerogen analysis very low (35.35%) and data should not be used in kerogen studies

Sample number: Stratigraphic unit:

K-8

Well name:

· Mississippian, Banff Fm. Shell BA Whitelaw No. 1

Location:

2-14-82-2-W6M

Depth (m): 1964.7-1971,4

1986.1-1990.0

Description:

medium grey slightly calcareous shale

Depositional environment:

marine

MAJOR OXIDES (%)		ASSOCIATED KEROGEN		
SiO <sub>2</sub>	51	Normalized composition (	SUM CHONS = 100)	
TiO <sub>2</sub>	0.4	Adjusted for pyrite sulfur	(% Fe x 1.1482)	
Al <sub>2</sub> O <sub>3</sub>	10			
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	3.9			
MgO	1.9	С	81.39	
CaO	2.3	Н	7.68	
Na₂O	0.0	0	7.58	
	Σ 69.5	N	3.14	
TRACE ELEMENTS	S (ppm)	S	0.21	
Ag	•			
Cd	12	Atomic H/C	1.124	
Co	19	O/C	0.0699	
Cr	116	N/C	0.0329	
Cu	26	s/c	0.0010	
Mn	120			
Мо	24	ESR properties (in vacuo,	corrected for pyrite)	
Ni	150			
Pb	10	spins/g Narrow	2.0x10 <sup>16</sup>	
Sr	46	Broad	2.0x10 <sup>18</sup>	
V	165	g-value Narrow	2.00283	
Zn	156	Broad	2.00311	
		width (mT) Narrow	0.10	
<u>ORGANICS</u>		Broad	0.67	
Organic C (%)	2.744			
Extractable		SULFUR ISOTOPES (δ345	S ‰ CDM)	
organics (ppm)	4140			
R <sub>e</sub> (average)	-	Kerogen	-15.2	
		Extractable organics	-5.8	
* = Below detection		Pyrite	-16.6	

## **COMMENTS**

- = Not determined

Sample number:

HE 32/35

Stratigraphic unit:

Mississippian, Exshaw Fm.

Well name:

Home Oil et al. C and E Elkton 7-3

Location:

7-3-31-4-W5M

Description:

very dark grey to black shale, with (?) conodonts

Depth (m): 3118.4-3120.3

Depositional environment: marine (euxinic)

- p		····a····o (ouxiiio)				
MAJOR OXIDES	<u>(%)</u>		ASSOCIATED KEROGEN			
SiO <sub>2</sub>	47		Normalized composition (SUM CHONS = 100)			
TiO <sub>2</sub>	0.3		Adjusted for pyrite sulfe	ır (% Fe x 1.1482)		
Al <sub>2</sub> O <sub>3</sub>	8					
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	2.0					
MgO	2.2		С	88.42		
CaO	>9.2		Н	4.39		
Na₂O	0.0		0	2.98		
	Σ >68.7		N	2.75		
TRACE ELEMEN	ITS (ppm)		S	1.46		
Ag	3.7					
Cd	10		Atomic H/C	0.592		
Co	•		O/C	0.0253		
Cr	126		N/C	0.0265		
Cu	29		S/C	0.0062		
Mn	115					
Мо	12		ESR properties (in vacu	o, corrected for pyrite)		
Ni	195					
Pb	•		spins/g Narrow	1.1x10 <sup>16</sup>		
Sr	352		Broad	1.9x10 <sup>19</sup>		
V	421		g-value Narrow	2.00265		
Zn	155		Broad	2.00305		
			width (mT) Narrow	0.098		
ORGANICS			Broad	0.61		
Organic C (%)	2.748					
Extractable			SULFUR ISOTOPES (8	<sup>34</sup> S ‰ CDM)		
organics (ppm)	3380					
R <sub>e</sub> (average)	0.66		Kerogen	+10.3		
			Extractable organics	+1.7		
* = Below detection	on		Pyrite	+6.7		

## **COMMENTS**

- = Not determined

Sample number: Stratigraphic unit: Well name:

HE 36

U.Devonian, Wabamun Gp. Home Oil et al. C and E Elkton 7-3

Location:

7-3-31-4-W5M

Description:

dark grey shale

Depositional environment:

marine

Depth (m): 3120.5-3123.0

MAJOR OXIDES (%)		ASSOCIATED KEROGEN			
SiO <sub>2</sub>	56	Normalized composition (	SUM CHONS = 100)		
TiO <sub>2</sub>	0.3	Adjusted for pyrite sulfur (	% Fe x 1.1482)		
Al <sub>2</sub> O <sub>3</sub>	11				
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	4.4				
MgO	2.5	С	87.98		
CaO	9.1	Н	4.16		
Na <sub>2</sub> O	0.1	0	3.77		
	Σ 83.4	N	2.73		
TRACE ELEMENTS	(ppm)	S	1.36		
Ag	•				
Cd	14	Atomic H/C	0.563		
Co	2	O/C	0.0322		
Cr	93	N/C	0.0265		
Cu	29	S/C	0.0058		
Mn	130				
Мо	20	ESR properties (in vacuo,	corrected for pyrite)		
Ni	159				
Pb	•	spins/g Narrow	-		
Sr	162	Broad	1.9x10 <sup>19</sup>		
V	71	g-value Narrow	2.00262		
Zn	100	Broad	2.00305		
		width (mT) Narrow	0.05		
ORGANICS		Broad	0.59		
Organic C (%)	2.792				
Extractable		SULFUR ISOTOPES (834S	% CDM)		
organics (ppm)	1540				
R <sub>e</sub> (average)	0.67	Kerogen	-10.7		
		Extractable organics	-5.6		
* = Below detection		Pyrite	-6.0		
- = Not determined					

## COMMENTS

Sample number: K-9 Stratigraphic unit: U.Devonian, Woodbend Gp., Ireton Fm. (middle unit) Well name: Socony Vegreville No. 1 Location: 14-20-51-15-W4M Depth (m): 943.7-963.2 Description: green-grey, calcareous shale Depositional environment: marine MAJOR OXIDES (%) ASSOCIATED KEROGEN SiO<sub>2</sub> 39 Normalized composition (SUM CHONS = 100) TiO, 0.3 Adjusted for pyrite sulfur (% Fe x 1.1482) Al<sub>2</sub>O<sub>3</sub> 9 3.3 ΣFe as Fe<sub>2</sub>O<sub>3</sub> MgO 2.6 C CaO 3.2 н Na<sub>2</sub>O 0.5 0 Σ 57.9 Ν TRACE ELEMENTS (ppm) s Ag 2.4 Cd 14 Atomic H/C Co 16 O/C Cr 104 N/C Cu 23 S/C Mn 394 Мо ESR properties (in vacuo, corrected for pyrite) Ni 130 РЬ spins/g Narrow Sr 279 **Broad** ٧ 38 g-value Narrow Zn 60 Broad width (mT) Narrow **ORGANICS Broad** Organic C (%) 0.180

# \* = Below detection - = Not determined

organics (ppm)

275

#### COMMENTS

Extractable

R<sub>a</sub> (average)

For the composition and mineralogy of the core from this well see Campbell and Oliver (1968).

Kerogen

**Pyrite** 

Extractable organics

SULFUR ISOTOPES (834S % CDM)

Sample number: K-10 Stratigraphic unit: U.Devonian, Woodbend Gp., Ireton Fm. (upper unit) Well name: Texaco Wizard Lake A-1 Location: 14-14-48-27-W4M Depth (m): 1889.8-1899.2 Description: green-grey, calcareous shale Depositional environment: marine MAJOR OXIDES (%) ASSOCIATED KEROGEN SiO<sub>2</sub> 60 Normalized composition (SUM CHONS = 100) TiO<sub>2</sub> 0.6 Adjusted for pyrite sulfur (% Fe x 1.1482) Al<sub>2</sub>O<sub>3</sub> 16 ΣFe as Fe<sub>2</sub>O<sub>3</sub> 5.0 MgO C 3.3 CaO н 3.2 0.1 0 Na<sub>2</sub>O Σ 88.2 N TRACE ELEMENTS (ppm) \$ Ag 16 Cd 16 Atomic H/C Co 48 O/C Cr 149 N/C Cu 23 S/C Mn 634 Мо 54 ESR properties (in vacuo, corrected for pyrite) Ni 150 РЬ spins/g Narrow Sr 658 Broad ٧ 101 g-value Narrow 77 Zn **Broad** width (mT) Narrow **ORGANICS Broad** Organic C (%) 0.204 Extractable SULFUR ISOTOPES (824S % CDM) organics (ppm) 280 R<sub>o</sub> (average) Kerogen

#### **COMMENTS**

\* = Below detection

- = Not determined

For the composistion and mineralogy of the core from this well see Campbell and Oliver (1968).

Extractable organics

Pyrite

Sample number:

K-11

Stratigraphic unit:

U.Devonian, Woodbend Gp., Majeau Lake Fm. Imp Westlock 14-24-59-26

Well name: Location:

14-24-59-26-W4M

Depth(m) 1562.4-1567.9

Description:
Depositional environment:

black, bituminous, slightly calcareous shale marine, close to carbonate reef complex

Doposidonal citti	J	maine, close to carbonate reer complex		
MAJOR OXIDES (	<u>%)</u>	ASSOCIATED KEROS	EN	
SiO <sub>2</sub>	56	Normalized compositio	n (SUM CHONS = 100	)
TiO <sub>2</sub>	0.4	Adjusted for pyrite sulf	ur (% Fe x 1.1482)	
Al <sub>2</sub> O <sub>3</sub>	12			
ΣFe as Fe <sub>z</sub> O <sub>3</sub>	3.6			
MgO	1,3	c	81.92	
CaO	3.4	н	8.43	
Na <sub>2</sub> O	0.0	0	6.36	
	Σ76.7	N	2.94	
TRACE ELEMENT	S (ppm)	S	0.35	
Ag	2.8			
Cd	2.1	Atomic H/C	1.226	
Co	39	O/C	0.0583	
Cr	101	N/C	0.0306	
Cu	7	s/c	0.0016	
Mn	204			
Мо	18	ESR properties (in vac	uo, corrected for pyrite)	)
Ni	70			
Pb	•	spins/g Narrow		
Sr	182	Broad	1.2x10 <sup>16</sup>	
V	84	g-value Narrow	2.00295	
Zn	46	Broad	2.00340	
		width (mT) Narrow	-	
<b>ORGANICS</b>		Broad	0.67	
Organic C (%)	5.764			
Extractable		SULFUR ISOTOPES (	5 <sup>34</sup> S ‰ CDM)	
organics (ppm)	36,600			
R <sub>o</sub> (average)	0.33	Kerogen	+7.4	
		Extractable organics	+13.9	
* = Below detection	1	Pyrite	+11.8	

#### **COMMENTS**

- = Not determined

For a description of this type section see Pollock (1967)

Sample number:

K-12

Stratigraphic unit: Welt name:

U.Devonian, Beaverhill Lake Gp. Home Regent "B" Swan Hills 4-4-67-11 4-4-67-11-W5M

Location:

Depth (m): 2805.4-2809.2

Description:

medium grey calcareous shale

Depositional environment:

marine

MAJOR OXIDES (%	<u>6)</u>	ASSOCIATED KEROG	<u>EN</u>	
SiO <sub>2</sub>	66	Normalized composition	(SUM CHONS = 100)	
TiO <sub>2</sub>	0.1	Adjusted for pyrite sulfu	ır (% Fe x 1.1482)	
$Al_2O_3$	3			
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	0.4			
MgO	0.7	С	89.46	
CaO	>9.2	н	5.64	
Na₂O	0.1	0	3.21	
	Σ >79.5	N	1.69	
TRACE ELEMENTS	G (ppm)	S	0.00	
Ag	•			
Cd	3.2	Atomic H/C	0.751	
Co	•	O/C	0.0269	
Cr	44	N/C	0.0161	
Cu	10	S/C		
Mn	70			
Мо	10	ESR properties (in vacu	o, corrected for pyrite)	
Ni	48			
Pb	•	spins/g Narrow	-	
Sr	200	Broad	1.8x10 <sup>19</sup>	
V	35	g-value Narrow	-	
Zn	19	Broad	2.00306	
		width (mT) Narrow	-	
<u>ORGANICS</u>		Broad	0.62	
Organic C (%)	0.684			
Extractable		SULFUR ISOTOPES (8	<sup>34</sup> S ‰ CDM)	
organics (ppm)	2120			
R <sub>e</sub> (average)	0.68	Kerogen	-15.1	
		Extractable organics	+12.3	
* = Below detection		Pyrite	-9.5	(machine unstable)

## COMMENTS

- = Not determined

Sample number:

K-13

Stratigraphic unit:

M.Devonian, U. Elk Point Gp., Watt Mountain Fm.

Gilwood Ss.

Well name:

IOE Mitsue 4-4G1-73-4

Location:

4-4-73-4-W5M

Depth (m): 1621.5-1623.4

Description:
Depositional environment:

near-shore, deltaic

MAJOR OXIDES	%)	ASSOCIATED KEROGE	EN .
SiO <sub>2</sub>	32	Normalized composition	<del></del>
TiO <sub>2</sub>	0.3	Adjusted for pyrite sulfur	•
Al <sub>2</sub> O <sub>3</sub>	12		•
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	3.0		
MgO	1.7	С	85.08
CaO	2.2	н	5.24
Na₂O	0.3	0	6.06
	Σ 51.5	N	2.61
TRACE ELEMENT	S (ppm)	s	1.01
Ag	2.1		
Cd	1.1	Atomic H/C	0.734
Co	16	O/C	0.0535
Cr	88	N/C	0.0262
Cu	10	S/C	0.0044
Mn	68		
Мо	12	ESR properties (in vacue	o, corrected for pyrite)
Ni	350		
Pb	•	spins/g Narrow	-
Sr	101	Broad	9.1x10 <sup>16</sup>
V	97	g-value Narrow	-
Zn	69	Broad	2.00291
		width (mT) Narrow	-
ORGANICS		Broad	0.42
Organic C (%)	0.208		
Extractable		SULFUR ISOTOPES (δ³	'S ‰ CDM)
organics (ppm)	184		
R <sub>e</sub> (average)	-	Kerogen	-
		Extractable organics	-
* = Below detection		Pyrite	-
- = Not determined			

## **COMMENTS**

For a description of this core see Kramers and Lerbekmo (1967). Composite sample taken over major shale bed (#262-264 of Kramers and Lerbekmo (1967))

Sample number: Stratigraphic unit: Well name:

K-14

M.Devonian, U.Elk Point Gp., Muskeg Fm. Banff Aquit Rainbow West 7-32-109-8 7-32-109-8-W6M

Location:

Depth (m): 1765.1-1765.4

Description: Depositional environment:

marine

Depositional envir	ronment:	marine		
MAJOR OXIDES	(%)		ASSOCIATED KEROGI	<u>EN</u>
SiO <sub>2</sub>	60		Normalized composition	(SUM CHONS = 100)
TiO <sub>2</sub>	0.9		Adjusted for pyrite sulfu	r (% Fe x 1.1482)
Al <sub>2</sub> O <sub>3</sub>	16			
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	7.0			
MgO	2.3		С	-
CaO	2.9		Н	-
Na₂O	0.1		0	•
	Σ 89.2		N	•
TRACE ELEMEN	TS (ppm)	•	S	0.0
Ag	0.5			
Cd	16		Atomic H/C	-
Co	11		O/C	-
Cr	221		N/C	-
Си	17		S/C	
Mn	85			
Мо	13		ESR properties (in vacu	o, corrected for pyrite)
Ni	260			
Pb	•		spins/g Narrow	2
Sr	10		Broad	2.1x10 <sup>16</sup>
V	51		g-value Narrow	5
Zn	54		Broad	2.00322
			width (mT) Narrow	-
ORGANICS			Broad	0.46
Organic C (%)	0.560			
Extractable			SULFUR ISOTOPES (δ3	<sup>4</sup> S ‰ CDM)
organics (ppm)	1120			
R <sub>o</sub> (average)	-		Kerogen	-
			Extractable organics	-
* = Below detection	n		Pyrite	-
- = Not determined	d d			

## COMMENTS

Sample number: Stratigraphic uni Well name:		K-16 M.Devonian, Nahanni Fm. Amoco Beaver B-2 6-19-K			
Location: Description:		b-19-K, 94-N-16		Depth (m): 4125.8	
Depositional env	rironment:	marine			
MAJOR OXIDES	S (%)		ASSOCIATED KEROGE	<u> </u>	
SiO <sub>2</sub>	0.5		Normalized composition	(SUM CHONS = 100)	
TìO <sub>2</sub>	0.0		Adjusted for pyrite sulfur	r (% Fe x 1.1482)	
Al <sub>2</sub> O <sub>3</sub>	2.1				
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	0.2				
MgO	18.2		C	89.52	
CaO	>9.2		Н	2.17	
Na <sub>2</sub> O	0.0		0	1.96	
	Σ >30.2		N	1.04	
TRACE ELEMENTS (ppm)			S	5.31	
Ag	•				
Cd	18		Atomic H/C	0.289	
Co	13		O/C	0.0164	
Cr	233		N/C	0.0099	
Cu	9		S/C	0.0222	
Mn	144				
Mo	15		ESR properties (in vacua	o, corrected for pyrite)	
Ni	136				
Pb	•		spins/g Narrow	-	
Sr	27		Broad	5.4x10 <sup>16</sup>	
V	6		g-value Narrow	-	
Zn	30		Broad	2.00316	
			width (mT) Narrow	•	
ORGANICS			Broad	0.066	
Organic C (%)	-				
Extractable			SULFUR ISOTOPES (824S %. CDM)		
organics (ppm)	-				
R <sub>e</sub> (average)	-		Kerogen	+5.2	
			Extractable organics	-	
* = Below detecti	on		Pyrite	÷	

## COMMENTS

- = Not determined

Rare micrinite (?) and fusinite (R<sub>o</sub> ~ 2.5)

Sample number:

K-15

Stratigraphic unit:

- M.Cambrian, Stephen Fm. Imperial Leduc No. 530

Depth (m): 2653.3-2659.4

Well name: Location:

8-17-50-26-W4M

Description: Depositional environment:

marine

Doposiusitai siivii	Similarity mainta		
MAJOR OXIDES	<u>(%)</u>	ASSOCIATED KEROGEN	Ī
SiO <sub>2</sub>	62	Normalized composition (	SUM CHONS = 100)
TiO <sub>2</sub>	0.9	Adjusted for pyrite sulfur (	% Fe x 1.1482)
Al <sub>2</sub> O <sub>3</sub>	26		
ΣFe as Fe <sub>2</sub> O <sub>3</sub>	9.4		
MgO	3.1	C	-
CaO	0.3	Н	-
Na <sub>2</sub> O	0.1	0	-
	Σ 101.8	N	·
TRACE ELEMENT	TS (ppm)	S	-
Ag	•		
Cd	19	Atomic H/C	-
Co	11	O/C	•
Cr	275	N/C	-
Cu	60	S/C	-
Mn	129		
Мо	11	ESR properties (in vacuo,	corrected for pyrite)
Ni g	89		
Pb	•	spins/g Narrow	200
Sr	113	Broad	-
V	51	g-value Narrow	
Zn	129	Broad	
		width (mT) Narrow	-
<u>ORGANICS</u>		Broad	-
Organic C (%)	0.140	ni.	
Extractable		SULFUR ISOTOPES (δ34S	5 %. CDM)
organics (ppm)	117		
R <sub>e</sub> (average)		Kerogen	•
		Extractable organics	-
* = Below detectio	n	Pyrite	-

## **COMMENTS**

- = Not determined

For a description of the core from this well see Pugh (1971, Fig. 16)

## APPENDIX B

# Organic Petrography and Reflectance Measurements of Selected Alberta Shales

(Report by INRS)

ARC 'COAL' PETROGRAPHY AND REFLECTANCE (Ro) ON ISOLATED SAMPLES

Author: INRS-Pétrole/HY/lm 01.10.76 Addressee: Alberta Research Council

Reference: Isolated samples

## INTRODUCTION

Petrographic determination and reflectance measurement were performed on the organic matter of 15 isolated samples. Some difficulties were encountered due to the fact that the samples were assumed to be composed entirely of kerogen which was not the case. The problem was solved by a chloroform extraction. Technical problems related to the electronic stability of the reflectometer were solved and a new series of measures was obtained on the instrument.

## **RESULTS**

## A. ORGANIC MATTER PETROGRAPHY

<u>Samples K-1 to K3:</u> mainly composed of vitrinite-like matter associated with a high content of mineral matter. There are a few pyrofusinite particles with 2.00<Ro<3.75. Vitrinite-like matter: rare cells structure, without anisotropy or fluorescence. Grain size of 4-20 microns.

Samples K-4 and K-5: mainly mineral matter with ~30% of organic matter composed of

- ± 50% vitrinite (télenite)\*
- $\pm$  40% (huminite-exinite?) (Ro  $\approx$  0.35)
- ± 10% inertinite (semi-fusinite Ro ≈ 1.35) (fusinite 2.0<Ro<5.0)

Sample K-6: same microlithotype as K-5 but with highest content of mineral material (clay, quartz, pyrite). Maceral grain size smaller than that of K-5 (~ 10-20 microns).

Sample K-11: less than 2% of vitrinite and other macerals in argillaceous matter.

Sample K-12: vitrinite-like substance with ~ 10% of micron-size disseminated mineral matter.

Sample K16: rare micrinite?? and fusinite (Ro~2.5) in mineral matter.

<u>Samples BH:</u> trimacerite rich in mineral matter (pyrite and argillaceous material); <1% bituminite or exudatinite (yellow fluorescence) after chloroform washing; maceral grain size <100 microns with an average of 20 microns. Huminite red brown in transmitted light (0.12<Ro<0.30); Pyrofusinite (3.2<Ro<4.7); Semi-fusinite (1.1<Ro<1.7).

Samples RCA: organic matter similar to samples BH

#### B. REFLECTANCE VALUES

The reflectance values obtained are shown on Table 1.

<sup>\*</sup>Vitrinite is brownish in transmitted light.

Table 1. Reflectance values

Sample no.	Low values	Ro average	High values	Low values	Ro average	High values	Lectures	Low values	Ro average	High values
K-1	.84	1.27	1.74	.82	1.28	1.82	50	.82	1.27	1.82
K-2	.41	.82	1.55	.51	.72	.98	65	.41	.79	1.55
K-3	.29	.70	1.37	.53	.67	.95	65	.29	.70	1.37
K-4	.61	.90	1.37	.40	.77	1.20	65	.40	.88	1.37
K-5	.33	.49	1.17	.32	.50	.78	65	.32	.50	1.17
K-6	.27	.44	.75	.23	.51	.84	20	.23	.47	.84
K-11	.23	.29	.40	.26	.35	.53	20	.23	.33	.53
K-12		.65		.46	.68	.90	20	.46	.68	.90
K-16			No vit	rinite						
BH433-A	.27	.59	.96	.40	.65	.84	65	.27	.61	.96
BH431-G	.38	.70	1.03	.50	.60	.74	65	.38	.67	1.03
RCA65-1	.50	.72	.99	.41	.65	.85	65	.41	.70	.99
RCA67-1	.50	.68	.90	.50	.68	.84	65	.50	.68	.90
HE36				.36	.67	.93	50	.36	.67	.93
HE32-35				.39	.66	.96	50	.39	.66	.96
Reflectance values on telex dated Reflectance values obtained 76-09-20 after recalibration of instrument			V		ctance (in culated from columns	•				